

Enantioselective Separation of Atropisomeric PBB 132 and PBB 149 in Extracts from a Norwegian White-Tailed Sea Eagle Egg



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INTRODUCTION

Technical mixtures of polybrominated biphenyls (PBBs) have been extensively used as flameretardants in textile and electronic industries, and as additives in plastics¹. Despite a continuous reduction of the production in the last decades, the presence of PBBs in the environment was recently confirmed in a wide range of samples². Under environmental conditions, many PBB congeners form stable atropisomers. The enantiomer separation of atropisomeric PBBs isolated from a technical mixture was recently published³. The purpose of this work was to study the enantioselective fate of the environmentally relevant PBBs 132 (2,2',3,3',4,6'-hexabromobiphenyl) and 149 (2,2',3,4',5',6- hexabromobiphenyl, Figure 1) in egg extracts from Norwegian birds of prey.



Figure 1. Structures of atropisomers of PBB 149

EXPERIMENTAL

- Egg sample: A non-hatched white-tailed sea eagle egg from Vikna (Norway) collected in 1998. This egg contained 17 ng/g wet weight sum PBBs, whereof 10 ng/g wet weight PBB 153⁴
- Sample preparation: Homogenisation with Na₂SO₄, cold-column extraction, GPC, florisil column, group separation on silica column⁵, normal phase HPLC fractionation
- Enantioselective HPLC: Enantioseparation of PBB 132 on a column coated with heptakis(2,3,6-tri-O-methyl)-β-cyclodextrin on silica employing a flow of 0.5 mL/min acetonitrile/water (60:40, v/v)³. Quantitative analysis of HPLC fractions by non-chiral GC/EI-MS
- Enantioselective GC/EI-MSMS: Enantioseparation of PBB 149 on a column coated with 35 % randomly derivatized 6-O-tert.-butyl-dimethylsilyl-2,3-di-O-methyl-β-cyclodextrin in PS086 (β-TBDM)³. EI-MSMS detection: m/z 627.6 → m/z 546.7

Sample preparation

not detected

 Table 1. Separation of brominated compounds from PCBs on 8 g silica







Approximately 12 ng PBB 132 and 3 ng PBB 149 were enriched in two different fractions from 20 g egg sample (see Table 2)

RESULTS AND DISCUSSION Enantioselective separation of PBB 132 atropisomers



Figure 2. A) HPLC/UV enantioseparation of a PBB 132 reference standard isolated from the technical mixture. B) GC/EI-MS chromatograms of the five fractions obtained from enantioselective HPLC separation of the white-tailed sea eagle egg extract.

Enantiomeric ratio: Bird egg (Figure 2B) 0.92 - 0.99 (n=4)



Figure 3. Enantioselective GC/EI-MSMS separation of PBB 149 in the technical mixture Firemaster BP-6 $^{\odot}$ (A) and an extract of the white-tailed sea eagle egg (B).

Enantiomeric ratio: Bird egg (Figure 3B) 0.68 – 0.72 (n=3) Firemaster® (Fig. 3A) 0.97 – 1.02 (n=4)

CONCLUSIONS

- A very efficient clean-up of the egg sample was developed, leaving mainly PBBs in the extract.
- Atropisomers of PBB 132 could not be separated by enantioselective GC. A combination of enantioselective HPLC and non-chiral GC/MS quantification of the HPLC extracts proved to be successful. Deviation from the racemic mixture could not be found for PBB 132 atropisomers in the white-tailed sea eagle egg.
- Atropisomers of PBB 149 were successfully separated by enantioselective GC/EI-MSMS. An enantiomeric ratio of 0.7 and hence a deviation from the racemic ratio was found in the Norwegian bird of prey egg.

REFERENCES

- de Boer, J., de Boer, K., Boon, J.P. (2000) Polybrominated biphenyls and diphenylethers; in: The Handbook of Environmental Chemistry, Vol 3, Part K, Paasivirta J. Ed., Springer Verlag
 BFR (2001) The Second International Workshop on Brominated Flame Retardants, 14.-16.5.2001, Proceedings, The Swedish Environmental Protection Agency
- 3. Berger, U., Vetter, W., Götsch, A., Kallenborn, R. (2002) J. Chromatogr. A, 973, 123-133
- 4. Herzke, D., Berger, U., Kallenborn, R., Nygård, T., Vetter, W. Chemosphere, submitted
- 5. Krock, B., Vetter, W., Luckas, B. (1997) Chemosphere, 35, 1519-1530